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*Avogadro's Number, The Mole, Grams, Atoms, Molar Mass Calculations - Introduction*  
~~Chemquest 42 Mole Conversions Made Easy: How to Convert Between Grams and Moles~~ how many moles of  $\text{PbSO}_4$  can be produced from 24.0 g of  $\text{Pb}$  (and excess  $\text{Cl}_2$ ) GCSE Chemistry - The Mole (Higher Tier) #24 **Moles and Gases** 02 Moles and Gases GCSE Science Revision Chemistry  
"Calculating Moles of an Element" 1 - 9701\_w16\_qp\_23 : Moles and Stoichiometry, Moles and Concentration The Mole: Avogadro's Number and Stoichiometry

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Unit 10 Molar Volume and Mass

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Moles 3 - moles of GASStep by Step

*Stoichiometry Practice Problems | How to Pass*

*Chemistry How to Use a Mole to Mole Ratio |*

*How to Pass Chemistry Molarity Made Easy: How*

*to Calculate Molarity and Make Solutions*

**How to Solve Stoichiometry Problems with Chem in**

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**How many moles of CO<sub>2</sub> are produced when 2.0 moles of C<sub>3</sub>H<sub>8</sub>**

**react with an excess of O<sub>2</sub>? Remember, the**

**products of this reaction are CO<sub>2</sub> and H<sub>2</sub>O.**

**Interconverting Masses, Moles and Numbers of**

**Particles - Chemistry Tutorial**

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STOICHIOMETRY - Limiting Reactant \u0026

Excess Reactant Stoichiometry \u0026 Moles

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Limiting Reactant Practice Problem

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How to Balance a Chemical Equation EASY

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How to Calculate Molar Mass Practice Problems

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GCSE Science Revision Chemistry \"Calculating

Mass of a Number of Moles\"**Stoichiometry:**

**Moles to Moles - Practice - 1**

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Molar Volume Chap 8f: Mole| Relationship

between Mole \u0026 Molar mass |Numerical

~~Ideal Gas Calculations with Density and Molar~~

~~Mass: Chapter 10 - Part 3 Stoichiometry -~~

*Moles*

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Gas Stoichiometry: Equations Part 1

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Gas (# of moles of gas is constant) Gas # of moles of Gas in container (mol) Volume of Container (L) Temperature of Gas (K) Pressure of Gas (kPa) A 0.3008 14.0 672.0 120.0 B 0.3008 16.0 1280 200.0 C 0.3008 10.0 340.0 85.0 Table 2: Experimental Data for Three Different Samples of Gases That Contain Different Amounts of Gas (# of moles of gas is ...

## **Information: The Gas Constant**

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3. On planet X, 2.78 moles of a gas takes up 1.85 L under a pressure of 74.1 kPa and a temperature of 201oC. What is the value of the ideal gas constant (R) on planet X? (include units) 0.104 kPa-L/(mol-K) 4. At a pressure of 103 kPa and a temperature of

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22oC, 52.9 g of a certain gas has a volume of 31.5 L. What is the identity of this gas?

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36 Intro To Gases (Hint: find the molar mass  
of the gas and match it with the periodic  
table.) Molar mass = 39.97 g/mol ! Argon 5.  
Some oxygen gas has a volume of 41.0 L under  
a pressure of 245 kPa and a temperature of  
279 K.

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